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[Enthalpy Concentration Lithium Bromide Water](#)

Lithium hydroxide and hydrobromic acid (aqueous solution of hydrogen bromide) will precipitate lithium bromide in the presence of water. $\text{LiOH} + \text{HBr} \rightarrow \text{LiBr} + \text{H}_2\text{O}$ Uses. A 50-60% aqueous solution of lithium bromide is used in air-conditioning systems as desiccant.

[Standard enthalpy of formation - Wikipedia](#)

Carbon tetrabromide appears as a colorless crystalline solid. Much more dense than water and insoluble in water. Toxic by ingestion. Vapors are narcotic in high concentration. Used to make other chemicals.

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Diborane is a colorless gas with an offensive odor. It is toxic by inhalation. Diborane is decomposed by water giving off hydrogen, another flammable gas and boric acid a non-regulated material. Its vapors are heavier than air. Long term exposure to low concentrations or short term exposure to high concentrations can result in adverse health effects from inhalation.